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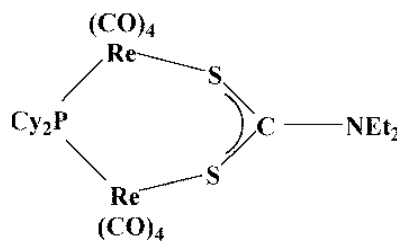
Key indicators

Single-crystal X-ray study
 $T = 203\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.018\text{ \AA}$
 R factor = 0.051
 wR factor = 0.149
Data-to-parameter ratio = 19.7For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.Octacarbonyl- μ -dicyclohexylphosphido-
 μ - N,N -diethyldithiocarbamate-dirhenium

The title compound, $[\text{Re}_2\{\mu\text{-S}_2\text{NC}(\text{Et}_2)\}(\mu\text{-PCy}_2)(\text{CO})_8]$ or $[\text{Re}_2(\text{C}_5\text{H}_{10}\text{NS}_2)(\text{C}_{12}\text{H}_{22}\text{P})(\text{CO})_8]$, is the first structurally characterized rhenium complex with a bridging dithiocarbamate ligand. The $\text{Re}\cdots\text{Re}$ distance is non-bonding.

Comment

The bromo bridge in $\text{Re}_2(\mu\text{-Br})(\mu\text{-PCy}_2)(\text{CO})_8$ is easily substituted by thiolate ligands giving phosphido-thiolato-bridged dirhenium complexes of the general formula $\text{Re}_2(\mu\text{-PCy}_2)(\mu\text{-SR})(\text{CO})_8$ (Egold *et al.*, 2001). We expected the former complex to react in a completely analogous manner with sodium diethyldithiocarbamate. However, this reaction gave the title compound, (I), in which, surprisingly, the dithiocarbamate ligand coordinates in a $\mu\text{-}\eta^1\text{-}\eta^1$ -mode.



(I)

The molecular structure of (I) shows a Re_2 unit which is doubly bridged by a $\mu\text{-PCy}_2$ group and the dithiocarbamate ligand. Each Re atom is additionally attached to four terminal CO ligands and attains thus a slightly distorted octahedral coordination. Carbonyl groups 4 and 7, which are both in *trans*-position to the $\text{Re}-\text{P}$ bonds, show eclipsed arrangement along the $\text{Re}\cdots\text{Re}$ vector with a $\text{C}-\text{Re}\cdots\text{Re}-\text{C}$ torsion angle of $-1.6(11)^\circ$, but the remaining CO ligands are staggered with torsion angles in the range $39.0(5)$ – $51.4(6)^\circ$. The clearly non-bonding $\text{Re}\cdots\text{Re}$ distance of $4.368(1)\text{ \AA}$ is bridged by the $\text{S}_2\text{C}(9)\text{NEt}_2$ ligand, with an $\text{S}\cdots\text{S}$ bite distance of $3.048(4)\text{ \AA}$. The accompanying $\text{S}-\text{C}-\text{S}$ angle is $123.1(7)^\circ$ and the $\text{S}-\text{Re}\cdots\text{Re}-\text{S}$ torsion angle of $57.8(1)^\circ$ shows the molecule to be twisted along the $\text{C}9\cdots\text{P}$ vector. The S_2CN group itself is planar and forms a dihedral angle of $47.3(2)^\circ$ with the Re_2P plane. The $\mu\text{-P}$ bridge is symmetric with equal $\text{Re}-\text{P}$ bond lengths of $2.562(3)$ and $2.561(3)\text{ \AA}$, whereas the two $\text{Re}-\text{S}$ bonds of $2.531(3)$ and $2.485(3)\text{ \AA}$ differ significantly. Similar patterns of likewise bridging dithio ligands are known for $(\text{Et}_4\text{N})[\text{Re}_2(\mu\text{-S})_2(\mu\text{-S}_2)(\mu\text{-S}_2\text{COR})(\text{S}_2\text{COR})_2]$ (McConnachie & Stiefel, 1999) and $\text{Re}_2(\text{CO})_8(\mu\text{-S}_2\text{CPCy}_3)$ (Alvarez *et al.*, 1991), both with direct $\text{Re}-\text{Re}$ bonds. The

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related Re—S bond lengths for these two compounds, of 2.442 (2)/2.462 (1) Å and 2.462 (5)/2.518 (5) Å, respectively, are somewhat shorter than those for (I). The first structure exhibits Re^{IV} atoms with strongly distorted octahedral S₆ coordination spheres and a completely planar Re₂μ-S₂CO fragment. For the second structure the same twisted M₂μ-S₂C geometry is observed as for (I), for Os₂N[S₂CN(CH₃)₂]₅ (Given & Pignolet, 1977) and for Ru₂(CO)₄(μ-S₂CH)(μ-PBu₂)(μ-dppm) (Böttcher *et al.*, 2000) as well. S··S bite distances of these other structures range from 3.02 to 3.15 Å and the S—C—S angles from 128.7 to 130.7°.

Experimental

In a Schlenk tube, 150 mg (0.172 mmol) Re₂(μ-Br)(μ-PCy₂)(CO)₈ was dissolved in 20 ml THF. 39 mg (0.228 mmol) Na₂CNET₂ was added and the reaction mixture was stirred for 24 h. The solvent was removed and the resulting residue subjected to TLC (eluent: dichloromethane/hexane, 1:10). From the major band, a pale yellow substance was isolated which, upon recrystallization from dichloromethane/pentane, gave 75 mg (46% yield) of the title compound. Spectroscopic data: ν(CO)-IR: 2090 (*vw*), 2077 (*s*), 1998 (*vs*), 1936 (*s*). ¹H NMR (CDCl₃): 1.23–2.34 (*m*, 28H; Cy, Me), 4.18 (*q*, ³J_{HH} = 6.8 Hz, 4H, NCH₂). ¹³C NMR (CDCl₃): 13.0 (*s*, Me), 26.0 [*s*, C⁴(Cy)], 27.5 [*d*, ²J_{CP} = 9 Hz, C²(Cy)], 33.2 [*s*, C³(Cy)], 41.0 [*d*, ¹J_{CP} = 11 Hz, C¹(Cy)], 51.2 (*s*, NCH₂), 183.5 [*d*, ²J_{CP} = 40 Hz, CO (*trans* μ-P)], 188.6 (*d*, ²J_{CP} = 6 Hz, CO); 190.2 (*d*, ²J_{CP} = 7 Hz, CO), 208.7 (*s*, S₂CN). ³¹P NMR (CDCl₃): 22.5 (*s*, μ-P).

Crystal data

[Re ₂ (C ₅ H ₁₀ NS ₂)(C ₁₂ H ₂₂ P)(CO) ₈]	Z = 2
M _r = 942.01	D _x = 2.050 Mg m ⁻³
Triclinic, P $\bar{1}$	Mo K α radiation
a = 9.218 (2) Å	Cell parameters from 25 reflections
b = 11.891 (2) Å	θ = 7.3–15.6°
c = 14.624 (2) Å	μ = 8.16 mm ⁻¹
α = 96.40 (1)°	T = 203 (2) K
β = 99.89 (2)°	Block, colorless
γ = 102.06 (2)°	0.30 × 0.26 × 0.20 mm
V = 1526.1 (5) Å ³	

Data collection

Bruker P4 diffractometer	R _{int} = 0.053
ω scans	θ_{\max} = 27.5°
Absorption correction: ψ scan (North <i>et al.</i> , 1968)	h = -11 → 1
T _{min} = 0.121, T _{max} = 0.187	k = -15 → 15
8235 measured reflections	l = -18 → 19
6945 independent reflections	3 standard reflections
5125 reflections with I > 2 σ (I)	every 397 reflections
	intensity decay: 2%

Refinement

Refinement on F ²	$w = 1/[\sigma^2(F_o^2) + (0.0703P)^2 + 15.7996P]$
R[F ² > 2 σ (F ²)] = 0.051	where P = (F _o ² + 2F _c ²)/3
wR(F ²) = 0.149	(Δ/σ) _{max} = 0.001
S = 1.05	$\Delta\rho_{\max}$ = 1.25 e Å ⁻³
6945 reflections	$\Delta\rho_{\min}$ = -1.04 e Å ⁻³
353 parameters	Extinction correction: SHELXL97
H-atom parameters constrained	Extinction coefficient: 0.0038 (4)

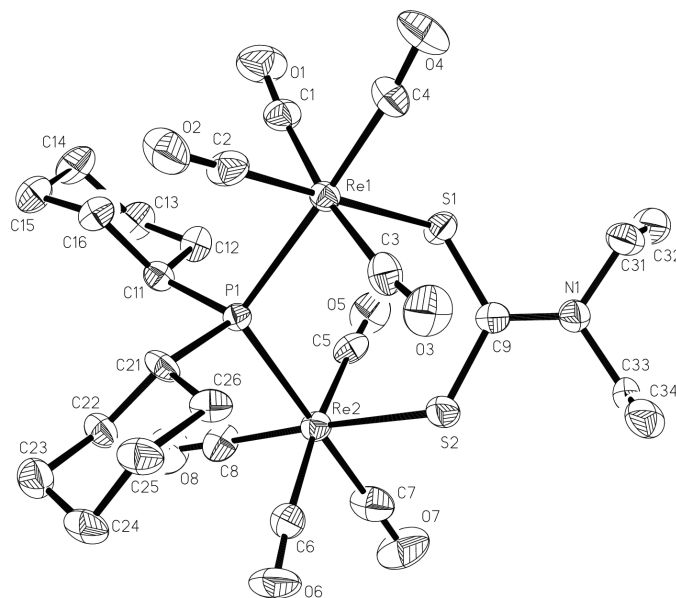


Figure 1

The molecular structure of (I) with H atoms omitted. Displacement ellipsoids are drawn at the 50% probability level.

Table 1

Selected geometric parameters (Å, °).

Re1—S1	2.531 (3)	Re2—P1	2.562 (3)
Re1—P1	2.561 (3)	S1—C9	1.740 (12)
Re2—S2	2.485 (3)	S2—C9	1.727 (12)
S1—Re1—P1	91.58 (10)	Re1—P1—Re2	117.03 (11)
S2—Re2—P1	91.86 (10)	N1—C9—S2	118.0 (9)
C9—S1—Re1	108.5 (4)	N1—C9—S1	118.9 (9)
C9—S2—Re2	114.9 (4)	S2—C9—S1	123.1 (7)

H atoms were placed at calculated positions, riding on the attached C atoms, with isotropic displacement parameters $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$, or $1.5U_{\text{eq}}(\text{C})$ for methyl groups. Residual electron density of 1.25 e Å⁻³ is found at (0.058, 0.871, 0.327), 0.96 Å from Re1.

Data collection: XSCANS (Bruker, 1996); cell refinement: XSCANS; data reduction: SHELXTL (Bruker, 1998); program(s) used to solve structure: SHELXTL; program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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